$$A \rightarrow B$$



To obtain some physical insights into the segregation model, consider drops falling into a reactor. There are three equally sized reactors per drop, each reactor will spend a different amount of time in the reactor. The reaction  $A \rightarrow B$  is occurring in each reactor



The drops will be spaced one minute apart as they leave the pipe on the way to the reactor. The reaction does not begin until the drop hits the reactor. To follow read up to t=3.



CD/Ch13p840insert.doc



Let's say the reaction  $A \rightarrow B$  is occurring in each of these little batch reactors. Because reactor 3 (*A*) will be in the reactor a longer time (3 min) it will have the highest conversion and reactor 1 (*C*) will be in the reactor the shortest time and have the lowest conversion. Lets say the conversion in reactor 1 is  $X_1$ =0.3, and reactor 2 is  $X_2$ =0.5, and reactor 3 is  $X_3$ =0.6. Each reactor represents one third of the total. The mean conversion is then

$$\overline{\mathbf{X}} = \frac{1}{3}(0.3) + \frac{1}{3}(0.5) + \frac{1}{3}(0.6) = 0.47$$